

LAMC Physical Science 1 Homework #5

This homework is due 10/20

Ch 5

Matching

c.14(radiation)

e.12(conduction)

f.1(temperature)

i.4(Kelvin Scale)

o.13(convective)

Multiple Choice

1.b

4.c

5.b

6.c

Ch 6

Matching

b.17(speed of sound)

c.10(electromagnetic spectrum)

d.5(amplitude)

f.8(period)

p.9(wave speed)

Multiple Choice

1.b

2.a

5.d

11.b

Fill in the blank

6.electromagnetic

7.longitudinal

13.natural

Ch 7

Matching

d.21(constructive interference)

f.6(refraction)

h.1(reflection)

u.3(law of reflection)

Multiple Choice

1.d

4.a

5.b

10.a

12.a

Ch 5.

Ex 6.

$$(a) T_c = T_k - 273$$

$$= 3 - 273$$

$$\boxed{= -270^\circ C}$$

(b)

$$T_F = (T_c + 40) \times 1.8 - 40$$

$$= (T_c \times 1.8) + 32$$

$$= (-270^\circ C + 40) \times 1.8 - 40$$

$$\boxed{= -454^\circ F}$$

Ex 14.

$$V = 1 \text{ L} \quad Q = m \cdot C \cdot \Delta T$$

First we need to find mass, m

$$\Delta T = 100^\circ C$$

$$C = 4186 \text{ J/kg}^\circ C$$

$m = \text{Volume} \times \text{density}$

$$= 1 \text{ L} \times 1 \text{ g/cm}^3, \text{ remember}$$

$$= 1000 \text{ cm}^3 \times 1 \text{ g/cm}^3$$

$$= 1000 \text{ g} = 1 \text{ kg}$$

* Specific Heat
can be found on page 112

From the previous homework!
we found that $1 \text{ L} = 1000 \text{ cm}^3$

$$Q = 1 \text{ kg} \times 4186 \text{ J/kg}^{\circ}\text{C} \times 100^{\circ}\text{C}$$

$= 418,600 \text{ J}$

$\Delta E_{\text{NC}} - T = \Delta T \quad (2)$

17.

$\Delta E_{\text{NC}} - \Delta E =$

$$m = 500 \text{ g}$$

$$\Delta T_1 = 10^{\circ}\text{C}$$

$$\Delta T_2 = 0^{\circ}\text{C}$$

$$\Delta T_3 = 20^{\circ}\text{C}$$

$$C_{\text{ice}} = 0.5 \text{ cal/g}^{\circ}\text{C}$$

$$C_{\text{water}} = 1 \text{ cal/g}^{\circ}\text{C}$$

$$L_f = 80 \text{ cal/g}$$

$$Q_1 = m C_{\text{ice}} \Delta T_1$$

$$= 500 \text{ g} \times 0.5 \text{ cal/g}^{\circ}\text{C} \times 10^{\circ}\text{C} = 2500 \text{ cal}$$

$$Q_2 = m L_f = 500 \text{ g} \times 80 \text{ cal/g} = 40,000 \text{ cal}$$

$$Q_3 = m C_{\text{water}} \Delta T_3$$

$$= 500 \text{ g} \times 1 \text{ cal/g}^{\circ}\text{C} \times 20^{\circ}\text{C} = 10,000 \text{ cal}$$

$$Q_{\text{Total}} = Q_1 + Q_2 + Q_3 = \boxed{52,500 \text{ cal}}$$

Ice at $-10^{\circ}\text{C} \rightarrow$ Ice at $0^{\circ}\text{C} \rightarrow$ Water at $0^{\circ}\text{C} \rightarrow$ Water at 20°C

ΔE_{NC} at 0°C ΔE_{NC} at 20°C ΔE_{NC} at 0°C

Q_1

Q_2

Q_3

$T \Delta S \Delta N = 0$

$1^{\circ}\text{C} = T \Delta V$

no zeroth law of thermodynamics

$1^{\circ}\text{C} = T \Delta P$

$\Delta H + \Delta S = 0$

$1^{\circ}\text{C} = T \Delta A$

~~$\Delta H = \Delta S \times T$~~

measured entropy at zero

$\Delta S(0) = 1 \text{ J/K}$ at zero

$\Delta T = \Delta S =$

Ch 6.

Ex 2. $f = 0.25 \text{ kHz}$

$$= 250 \text{ Hz}$$

$$T = \frac{1}{f}$$
$$= \frac{1}{250 \text{ Hz}}$$

$$\boxed{= 0.004 \text{ sec}}$$

Ex 14.

$$f = 15.0 \text{ kHz} \quad | \quad s = f \times \lambda$$

$$\lambda = 0.333 \text{ m.} \quad | \quad = 15000 \text{ Hz} \times 0.333 \text{ m}$$

$$\boxed{= 5000 \text{ m/s}}$$