

NOMENCLATURE PRACTICE

1	aluminum hydroxide	
2	iron(II) nitrate	
3	sodium iodide	
4	iron(III) sulfate	
5	aluminum sulfate	
6	strontium sulfide	
7	potassium nitrate	
8	tin(II) nitrate	
9	iron(II) bromide	
10	ammonium phosphate	
11	iron(III) fluoride	
12	sodium carbonate	
13	lithium nitride	
14	lithium carbonate	
15	sodium phosphide	
16	iron(III) oxide	
17	tin(IV) nitrate	
18	strontium nitrate	
19	ammonium carbonate	
20	sodium hydroxide	
21	iron(III) phosphate	
22	strontium bromide	
23	strontium phosphate	
24	aluminum carbonate	
25	ammonium phosphide	

26	potassium oxide	
27	strontium nitride	
28	ammonium hydroxide	
29	sodium sulfate	
30	aluminum fluoride	
31	aluminum sulfide	
32	strontium carbonate	
33	sodium bromide	
34	ammonium sulfate	
35	strontium iodide	
36	sodium oxide	
37	ammonium chloride	
38	lead(IV) oxide	
39	sodium nitride	
40	iron(II) fluoride	
41	strontium hydroxide	
42	ammonium sulfide	
43	lithium phosphate	
44	iron(II) iodide	
45	potassium hydroxide	
46	iron(III) iodide	
47	potassium sulfide	
48	strontium sulfate	
49	strontium oxide	
50	potassium fluoride	

Naming Covalent Compounds Worksheet

Write the formulas for the following covalent compounds:

- 1) antimony tribromide _____
- 2) hexaboron silicide _____
- 3) chlorine dioxide _____
- 4) hydrogen iodide _____
- 5) iodine pentafluoride _____
- 6) dinitrogen trioxide _____
- 7) ammonia _____
- 8) phosphorus triiodide _____

Write the names for the following covalent compounds:

- 9) P_4S_5 _____
- 10) O_2 _____
- 11) SeF_6 _____
- 12) Si_2Br_6 _____
- 13) SCl_4 _____
- 14) CH_4 _____
- 15) B_2Si _____
- 16) NF_3 _____