WHAT IS CHEMISTRY?

- Chemistry is the science that deals with the *materials* of the universe, and the *changes* they undergo.
- *Materials* of the universe can be of several forms:

Gas: air, oxygen

Liquid: water, gasoline, vinegar, orange juice,

Solid: rocks, charcoal, table salt, sugar, wood, baking soda

• Some examples of **changes**:

Burning of charcoal

 $charcoal + oxygen \longrightarrow carbon dioxide$

Burning of gasoline

gasoline + oxygen \longrightarrow carbon dioxide+ water vapor

Fermentation of grape juice

glucose \longrightarrow ethyl alcohol + carbon dioxide (in water) (in water)

Souring of wine

ethyl alcohol + oxygen \longrightarrow acetic acid (in water) (from air) (in water)

THE SCIENTIFIC METHOD

• The scientific method is a **process** of creative thinking and testing aimed at **objective** and **verifiable** discoveries. It is generally composed of the following steps:



Examples:

- 1. Which statement below describes a law? Which describes a theory?
 - a) It explains why observations occur.
 - b) A summary of previous observations that can be used to predict the results of future related experiments.
- 2. Classify each statement below as an observation, a law, or a theory:
 - a) All matter is made of small particles called atoms.
 - b) When iron rusts in a closed container, the mass of the container and its content does not change.
 - c) In a chemical reaction, matter is neither created nor destroyed.
 - d) When a match is burned, heat is released.
- 3. A student prepares several samples of the same gas and measures their mass and volume. The results are tabulated below:

Mass of Gas (in grams)	$\label{eq:Volume} \text{Volume of Gas (in } L)$	
22.5	2.5 1.60	
35.8	35.8 2.55	
70.2	5.00	
98.5	7.01	

Which statement(s) below describe the relationship between the mass and volume of the gas:

- a) The mass of the gas is inversely proportional to its volume.
- b) The mass to volume ratio of the gas is constant.
- c) The mass to volume ratio of the gas is not constant.
- d) The mass of the gas directly proportional to its volume.

Examples (cont'd):
4. Identify each of the following as an *observation*, a *hypothesis*, an *experiment* or a conclusion:

a)	Drinking coffee at night keeps me awake.	
b)	I will try drinking coffee only in the	
c)	morning. If I stop drinking coffee in the afternoon, I will be able to clean at night	
d)	will be able to sleep at night. When I drink decaffeinated coffee, I sleep	
u)	better at night.	
e)	I am going to drink only decaffeinated coffee.	
f)	I sleep better at night because I stopped drinking caffeinated drinks.	