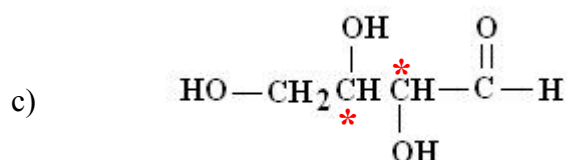
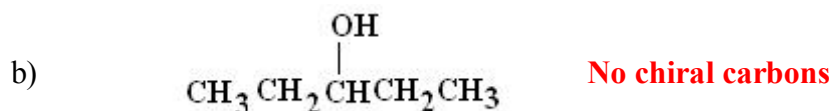
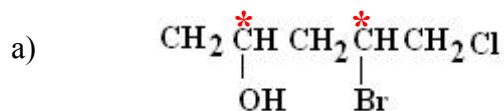


**REVIEW QUESTIONS**

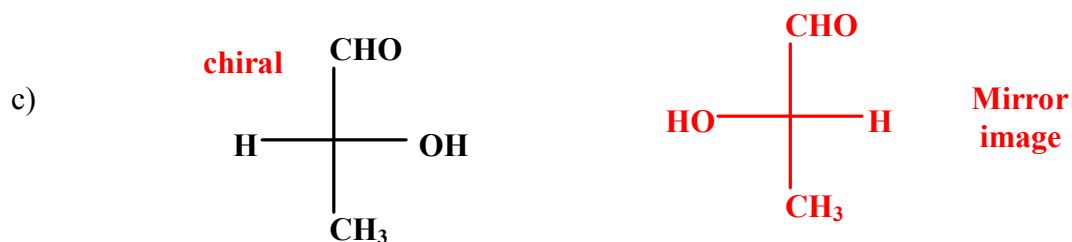
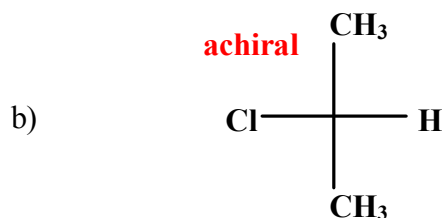
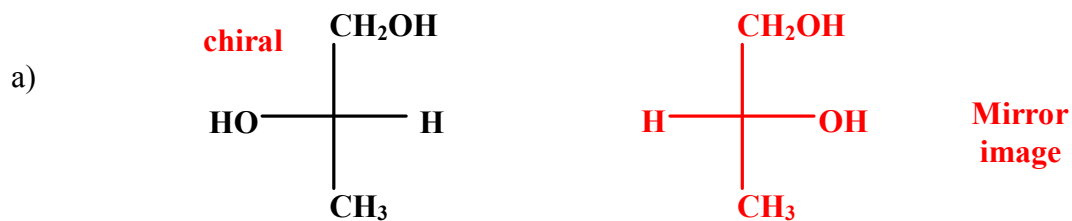
## Chapter 26

1. Identify any chiral carbon(s) in each compound shown below:

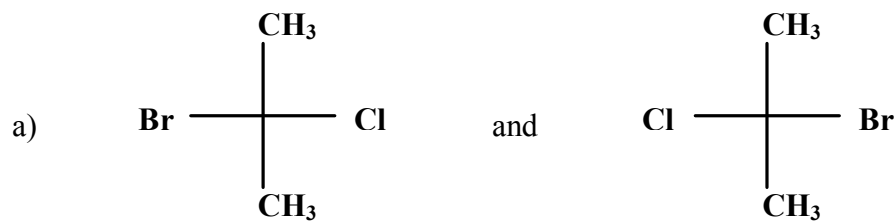
**Chiral carbons have 4 different groups attached to them, and are indicated by an asterisk (\*)**



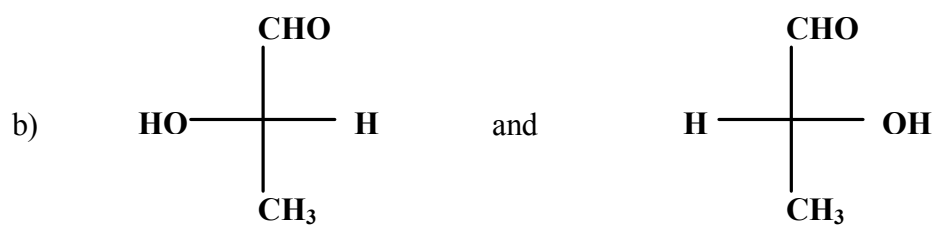
2. Determine if each of the Fisher projections shown below is a chiral compound. If so, draw the mirror image.



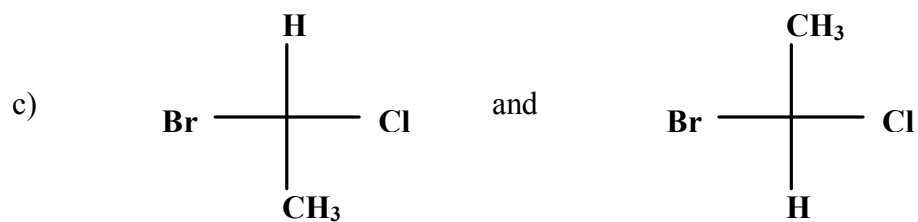
3. Indicate whether each pair of Fisher projections represent enantiomers or identical structures:



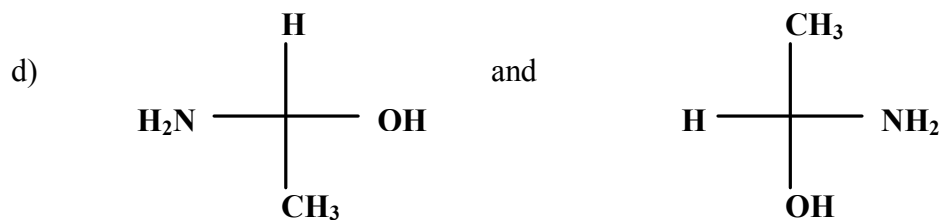
**Identical molecules since they are achiral**



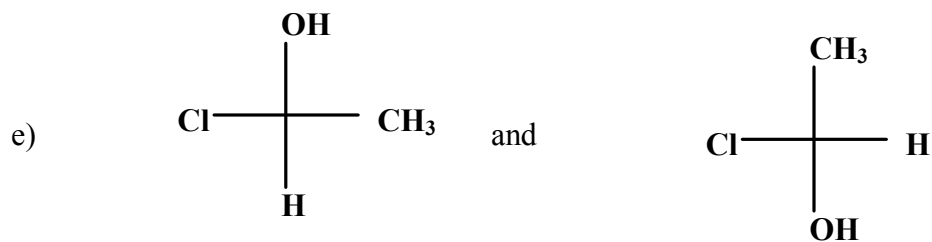
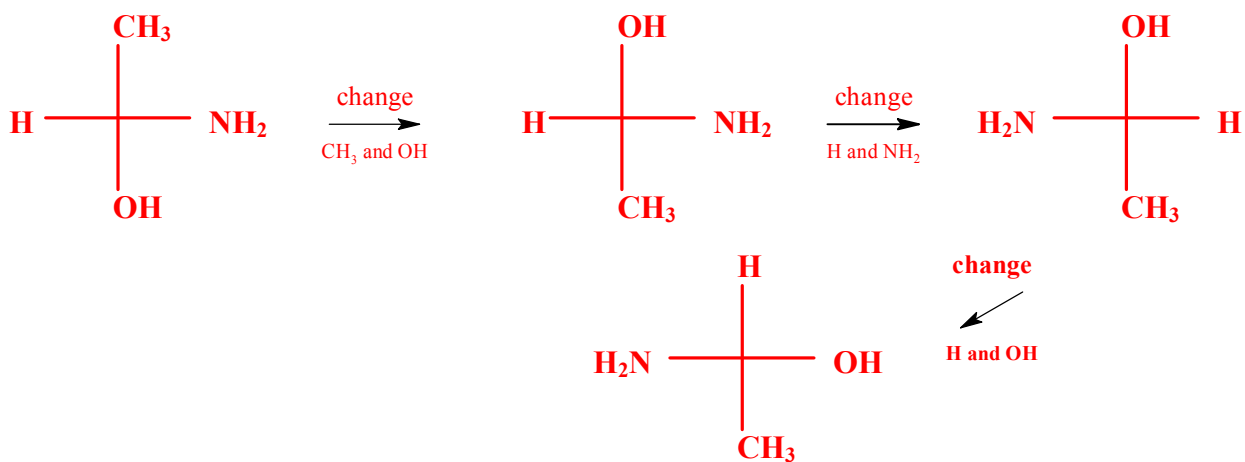
**Enantiomers since they are chiral and require one change (H and OH) to interconvert**



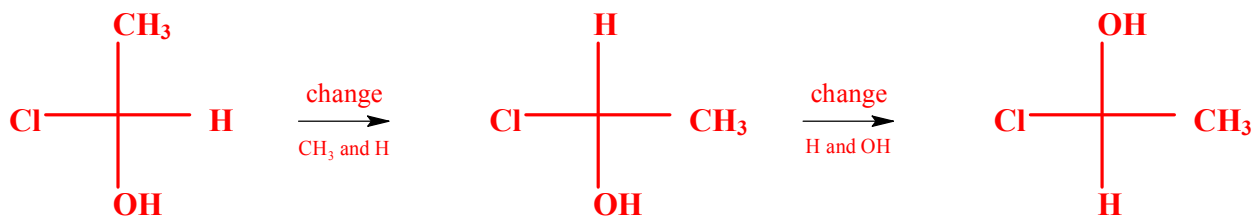
**Enantiomers since they are chiral and require one change (H and CH<sub>3</sub>) to interconvert**



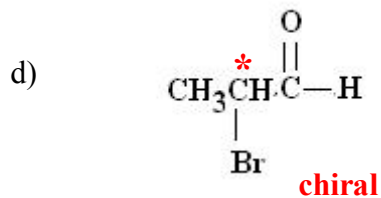
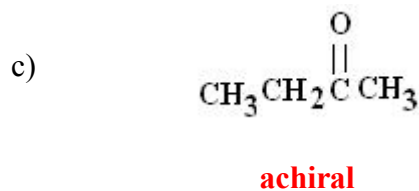
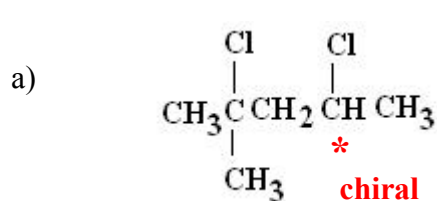
**Enantiomers since they are chiral and require 3 change to interconvert**



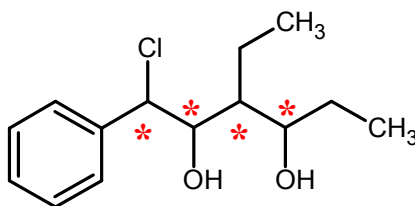
**Identical molecules since they are chiral and require 2 change to interconvert**



4. Identify each compound below as chiral or achiral. If chiral, indicate the chiral carbon. (**Chiral carbons are indicated with an asterisk**)

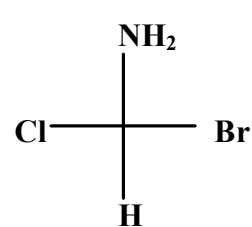
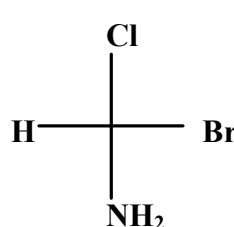
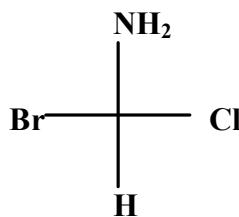
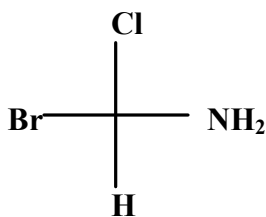


5. How many chiral carbons does the compound shown below possess? How many stereoisomers can this compound have?



**Four chiral carbons present (marked by asterisks)**  
**Total number of stereoisomers = 16 (2<sup>4</sup>)**

6. Three of the structures shown below are the same compound. The other is the enantiomer. Which is the enantiomer?



(a)

(b)

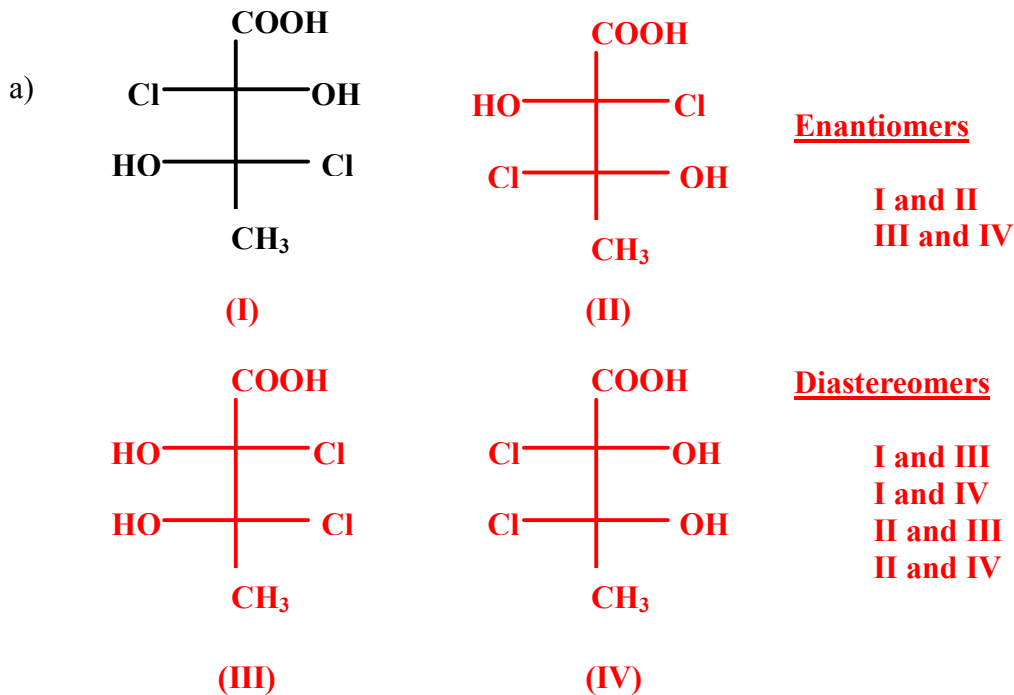
(c)

(d)

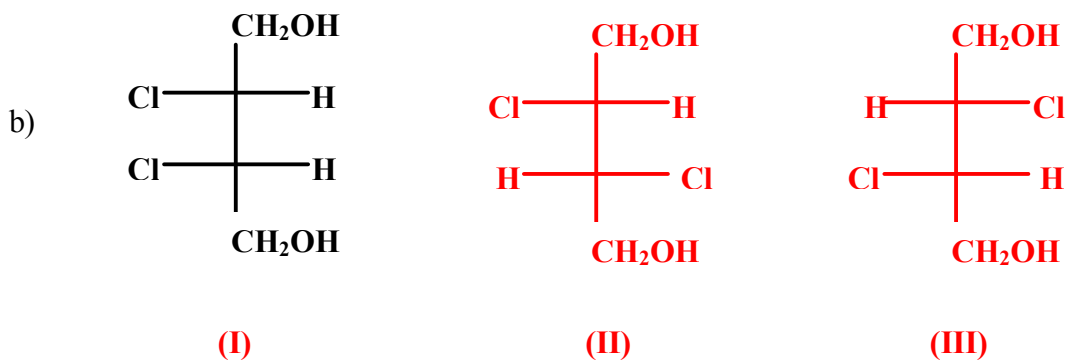
**Structure (b) is the enantiomer, since it requires one change (Cl and NH<sub>2</sub>) to interconvert to (a). Structures (c) and (d) are identical to (a) since each require 2 changes to interconvert.**

7. Draw all the stereoisomers possible for the structures below. Identify enantiomer pairs, diastereomer pairs and meso compounds.

**Since the molecule has 2 chiral carbons, there are 4 stereoisomers possible.**



**Since the molecule has 2 chiral carbons, there are 4 stereoisomers possible. However, due to the symmetry of the molecule one of the structures is meso.**



Enantiomers: II and III

Diastereomers: I and II  
I and III

Meso Compound: I