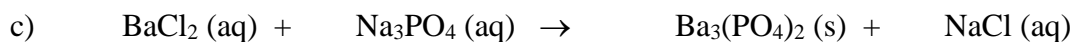
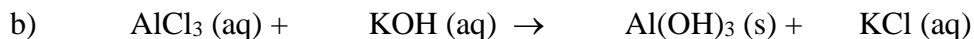
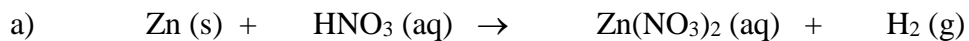


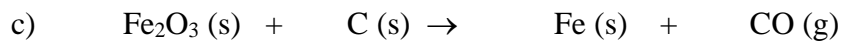
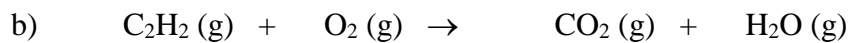
CHEMICAL REACTIONS & EQUATIONS**Exit Ticket 7**

- Write a balanced equation for each reaction described below. Include state designation for each reactant and product.
 - Dinitrogen oxide, laughing gas is produced by heating solid ammonium nitrate. The other product formed in this reaction is gaseous water.
 - When ethanol, C_2H_6O (aq), is consumed, it reacts with oxygen gas in the body to produce gaseous carbon dioxide and liquid water.

- Balance the following equations:



3. Balance and classify each of the following reactions as synthesis, decomposition, single replacement, double replacement or combustion:



4. Identify each of the following as oxidation or reduction, and complete by placing the proper number of electrons on the proper side of equation:

