

# Chemistry 51 Lab Experiments

*Experiment procedures, report forms and other documents for the laboratory experiments for this class are available below:*

<b><i>Experiment No.</i></b>	<b><i>Title</i></b>	<b><i>Report Forms</i></b>
1	<a href="#">Measurement</a>	<a href="#">Report Form</a>
2	<a href="#">Density</a>	<a href="#">Report Form</a>
3	<a href="#">Qualitative Separation of Mixtures</a>	<a href="#">Report Form</a>
4	<a href="#">Quantitative Separation of Mixtures</a>	<a href="#">Report Form</a>
6	<a href="#">Specific Heat of a Metal</a>	<a href="#">Report Form</a>
H/O	<a href="#">Nomenclature</a>	
H/O	<a href="#">Molecular Shape &amp; Polarity</a>	
H/O	<a href="#">Balancing Equations</a>	
10	<a href="#">Combination and Decomposition</a>	<a href="#">Report Form</a>
11	<a href="#">Single Replacement Reactions</a>	
12	<a href="#">Double Replacement Reactions</a>	<a href="#">Report Form</a>
H/O	<a href="#">Net Ionic Equations</a>	
H/O	<a href="#">Empirical Formula</a>	

<b><i>Experiment No.</i></b>	<b><i>Title</i></b>	
9	<a href="#"><u>Percentage of Copper in Malachite</u></a>	<a href="#"><u>Report Form</u></a>
13	<a href="#"><u>Table Salt from Baking Soda</u></a>	<a href="#"><u>Report Form</u></a>
H/O	<a href="#"><u>Electrolytes</u></a>	
H/O	<a href="#"><u>Properties of Acids &amp; Bases</u></a>	
H/O	<a href="#"><u>Charles' Law</u></a>	